Epitaxy

Mathews-Blakeslee Defect Thickness

$$h_{c} = \frac{b\left(1 - \frac{\nu}{4}\right)}{4\pi \left|\frac{a_{s} \cdot a_{1}}{a_{s}}\right|(1 + \nu)} \ln 1 + \frac{h_{c}}{b} \quad \nu \approx 0.3 \qquad b \approx 4 \text{ Å}$$

Chemical Vapor Deposition – Vapor Phase Epitaxy A few microns per hour

$$J = \frac{\beta P}{\sqrt{2\pi mk_B}}$$

$$K = \exp\left(\frac{\Delta G}{k_B T}\right)$$

$$R = \frac{c_s}{N} = \frac{c_g}{N} \left(\frac{h_g k_s}{h_g + k_s}\right)$$

$$R \propto \exp\left(-\frac{\Delta E}{k_B T}\right)$$

$$K = \frac{\beta P}{\sqrt{2\pi mk_B}}$$

$$K < 1, \text{ reactants favored}$$

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HUL

ScaAs(s)

Substrate

Schematic for the halide process.

750 °C Growth 900 °C Etch

Silicon SiH4 \rightarrow Si + 2 H2 SiH2Cl2 \rightarrow Si + 2 HCl B2H6, PH3, and AsH3 are used as dopants

HUG

Gallium

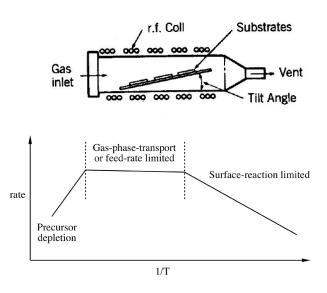
source 800-850 °C

From

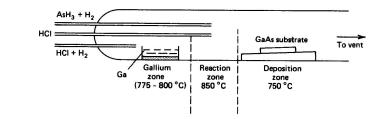
AsCI,

bubble

A,(S) ~ (~ (g)



GaAs Ga(CH2CH3)3, AsH3, In(CH3)3, Al(CH2CH3)3 H2S, H2Se, SiH4 are used as dopants

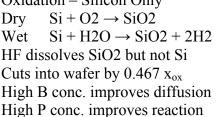


Schematic for the hydride process.

 $\begin{array}{ll} \text{SiH4} + \text{O2} \rightarrow \text{SiO2} \ 420\text{C} & \text{TEOS} + \text{O2} \rightarrow \text{SiO2} \ 720\text{C} & \text{SiH4} + \text{N2O} \rightarrow \text{SiO2} \ 275\text{C} \\ 2 \ \text{MoCl5} + 5 \ \text{H2} \rightarrow 2 \ \text{Mo} + 10 \ \text{HCl} & \text{Self-aligned contact: } 2 \ \text{WF6} + 3 \ \text{Si} \rightarrow 2\text{W} + 3 \ \text{SiF4} \\ \end{array}$

Vent

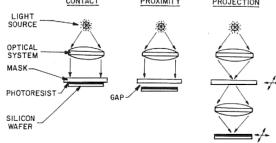
Physical Vapor Deposition – Molecular Beam Epitaxy $J = 1.12 \times 10^{22} \frac{PA_e \cos \theta}{l^2 \sqrt{M} \sqrt{T}} \quad \text{where M is molar mass}$ R = flux/crystal densityComposition is fractional pressure A few microns per hour Oxidation – Silicon Only

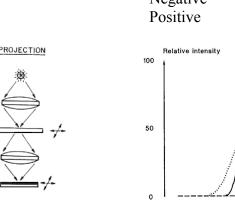


Slower, consistent ϵ Faster, variable ϵ effusion cell 3-zone furnace Wafers Water vapor or oxygen inlet Cap

Gas Concentration of Oxidant Molecules Oxide Silicon Deal-Grove Model $x_{ox}^2 + A x_{ox} = B (t + \tau)$ x0 A/2 $A = 2D\left(\frac{1}{k_s} + \frac{1}{h}\right) \quad B = \frac{2DC^*}{N} \quad \tau = \frac{x_i^2 + Ax_i}{B}$ Oxide C Thickness x₀ Surface Oxidant Chemica $\frac{B}{A} \approx C^* \frac{k_s}{N}$ if h>>k_s Reaction Diffusio Distance x $\frac{t+\tau}{A^2/4B}$ Lithography Photoresist Substrate ArF light is 193 nm Positive Negative developmen development $\sin \theta = \frac{1.22}{d} \lambda$ where d is aperture diameter Numerical aperture is n sin θ Feature size, $W_{\min} \approx \frac{0.6}{n \sin \theta} \lambda$ Polymer is more soluble Polymer is *less* soluble after exposure after exposure. light induces cross-linking Negative Depth of focus, $\sigma = \frac{1}{n^2 \sin^2 \theta} \lambda$ Positive lights breaks bonds Proximity CONTACT PROXIMITY PROJECTION Relative intensity 100 www Contact

FINAL FORMULA SHEET





Etches

Wet – there are many recipes

(111) is the slow plane for anisotropic wet etching in Si (KOH or TMAH)

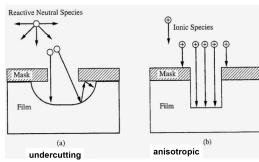
HNO3 then HF Fast etch 49% HF Through mask ~3 molar HF with NH4F Cleaning Dilute HF $2 \text{ GaAs} + 6 \text{ H2O} \rightarrow \text{Ga2O3} + \text{As2O3} + 12\text{H}^+$ Ga2O3 and As2O3 dissolve in acids and bases Caro's etch: H2SO4, H2O2, H2O (diffusion limited so flat surface

Dry and Reactive Ion Etching

Argon bombardment at 1 keV mills most materials at 10-100 nm/min Halocarbons etch through the radical formation of volatile products



Evaporation can non-selectively deposit metals without damage Sputtering (Ar) can dislodge substances which then shower the wafer Damascene: etch figure, fill with metal, planarize



Projection

SPENSER TALKINGTON